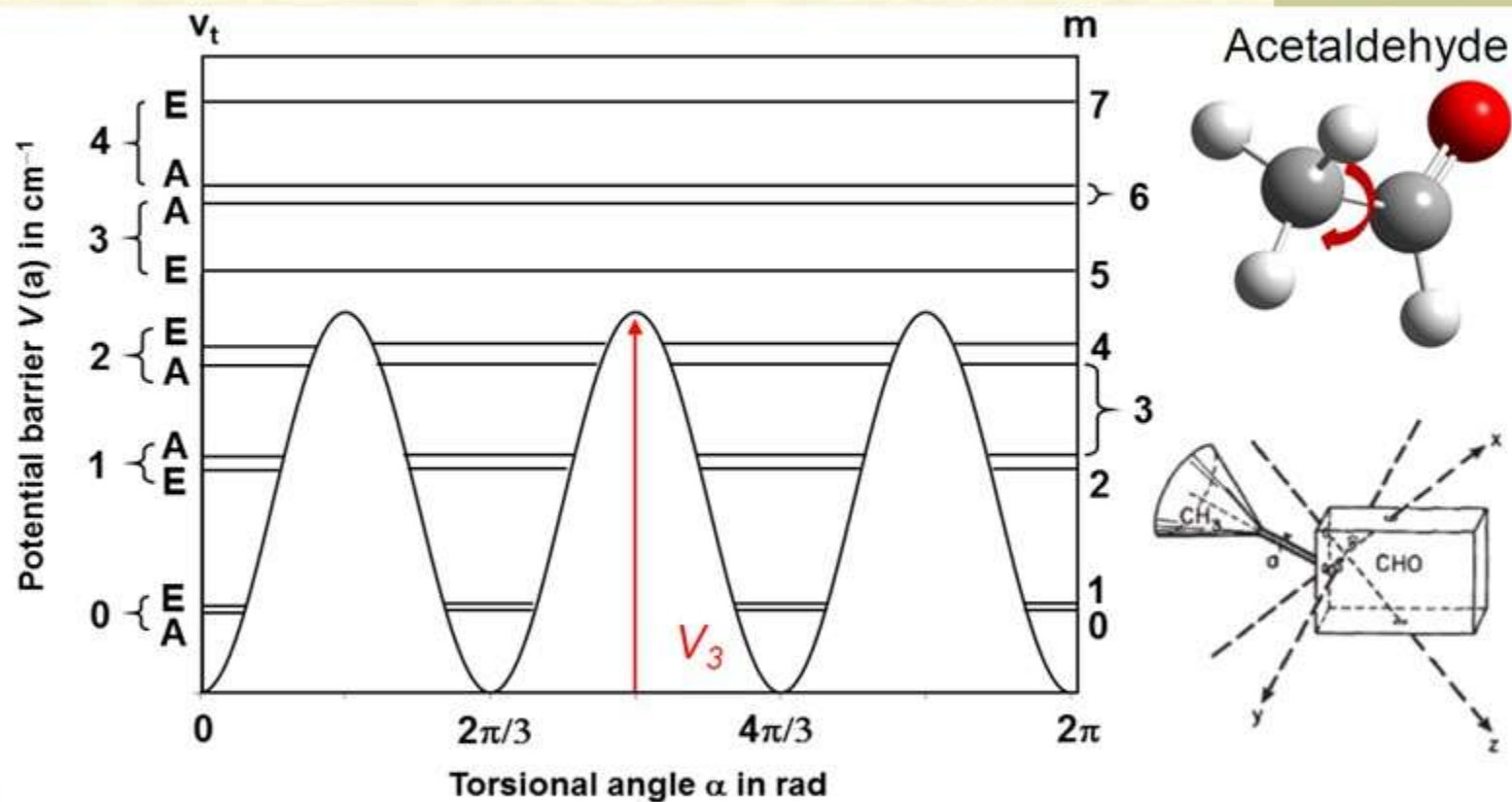


Internal rotation

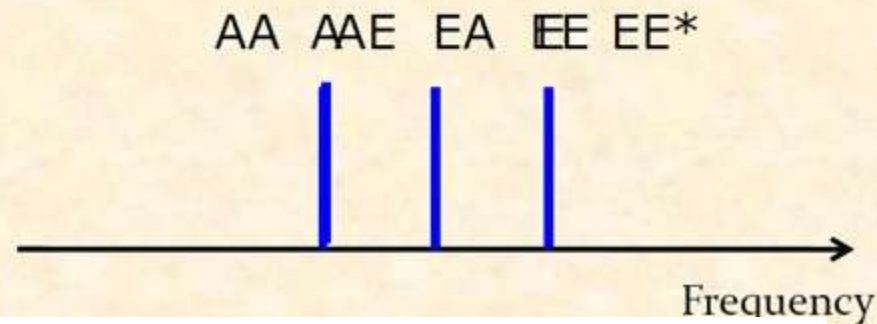
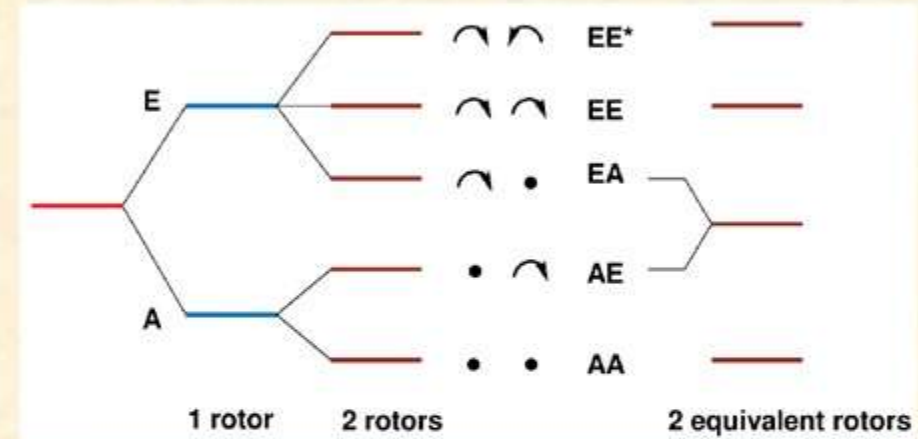
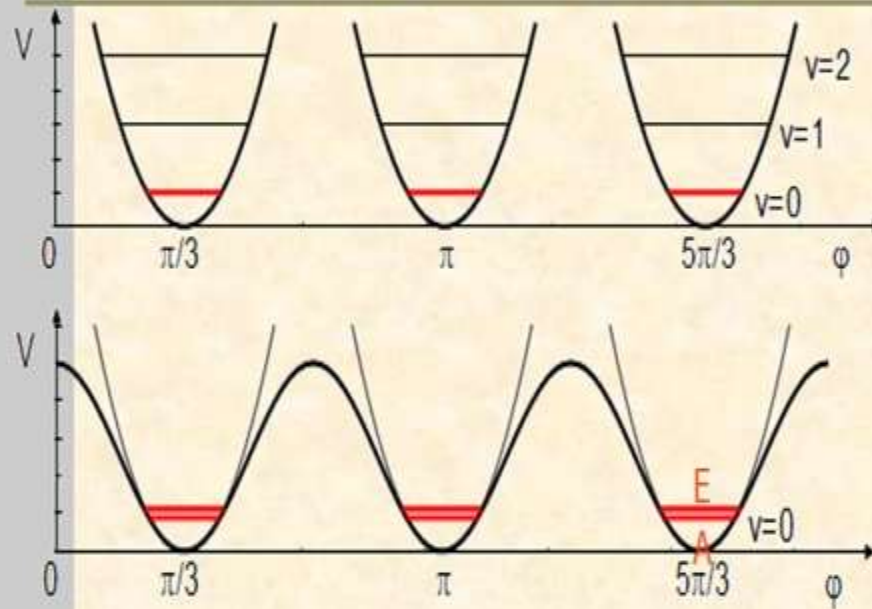


$$V(\alpha) = \frac{V_3}{2}(1 - \cos 3\alpha) + \frac{V_6}{2}(1 - \cos 6\alpha) + \dots$$

Lin and Swalen
Rev. Mod. Phys
1959

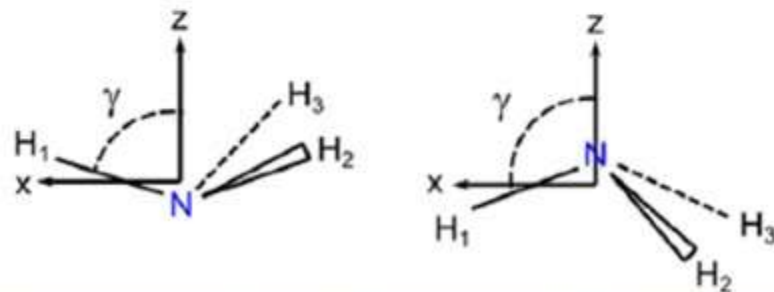
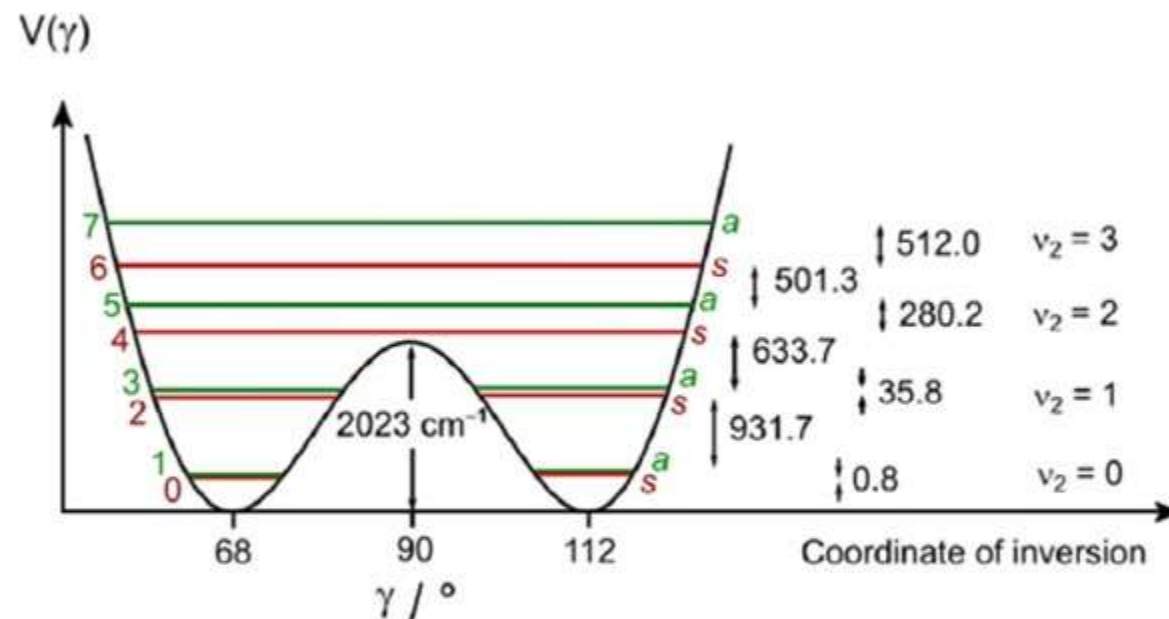
72

Internal rotation



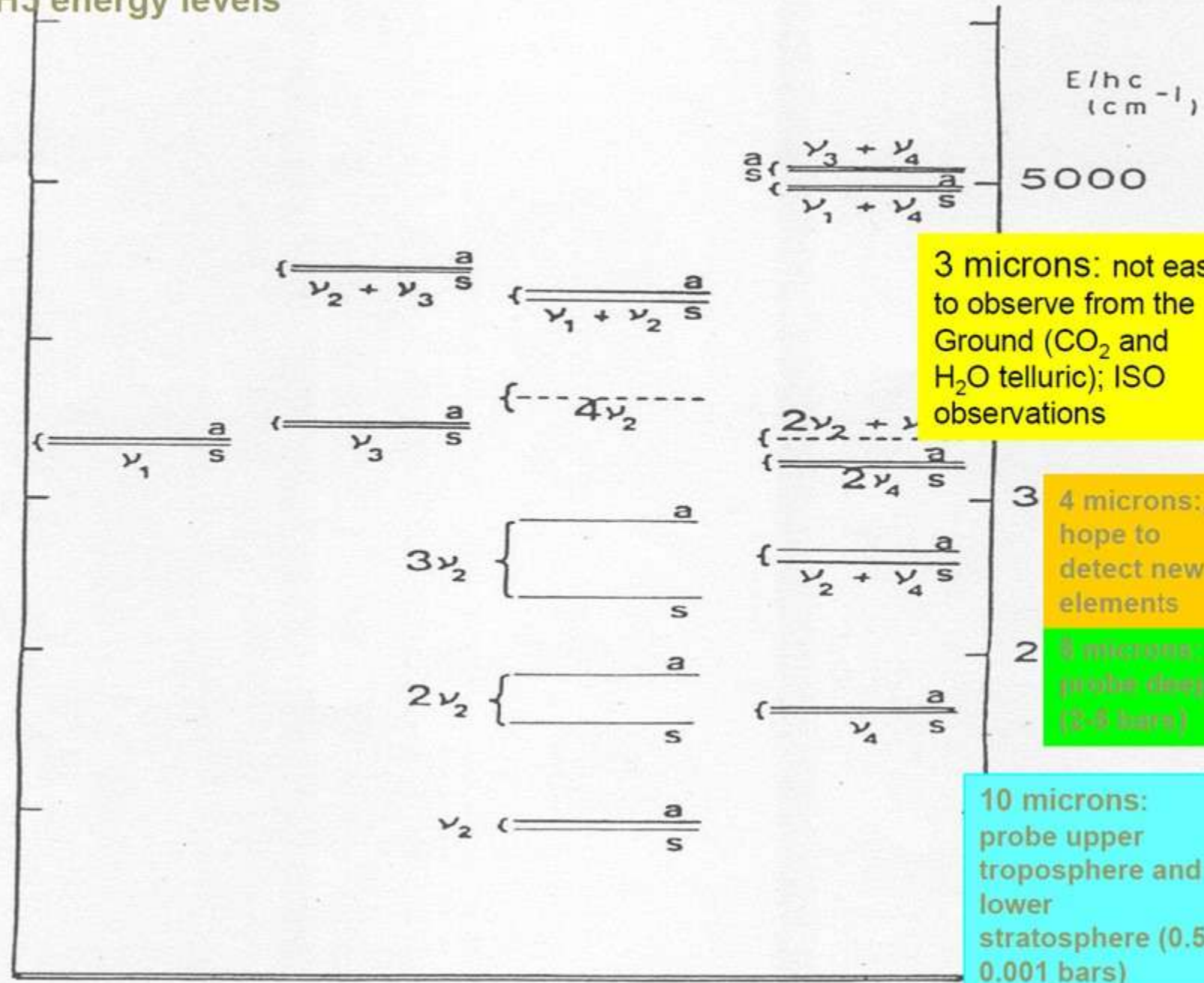
What makes NH_3 so hard???

Large amplitude motion : inversion



NH₃ ENERGY LEVELS

NH₃ energy levels



3 microns: not easy to observe from the Ground (CO₂ and H₂O telluric); ISO observations

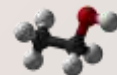
4 microns: hope to detect new elements

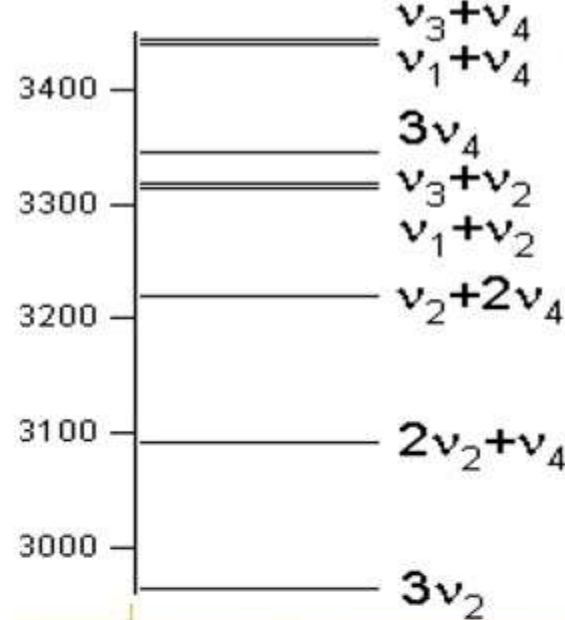
8 microns: probe deep (2-8 bars)

10 microns: probe upper troposphere and lower stratosphere (0.5-0.001 bars)

PH₃ energy levels

As NH₃, PH₃ presents an inversion energy level





-The band $3v_2$:

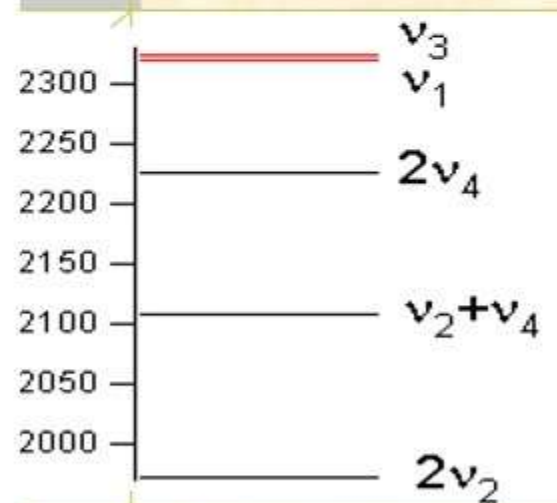
-Low resolution of line positions (Maki et al. JCP 1973)

- The **octad** bands:

New line positions and intensities measurements

(Butler et al., JMS 2006)

The Global analysis of the Dyad, pentad and octad bands: (Nikitin et al., JMS 2009)



• The $2v_2/v_2+v_4/2v_4/v_1/v_3$ pentad:

-Frequencies: rms= 0.009 cm^{-1} up to $J=16$

(Tarrago et al. JMS, 1992, Ulenikov and al. JMS, 2002)

-Intensities: rms= **13%** (Tarrago et al. JMS, 1992)



•The v_2/v_4 dyad:

-Frequencies: rms= 0.0004 cm^{-1} up to $J=22$

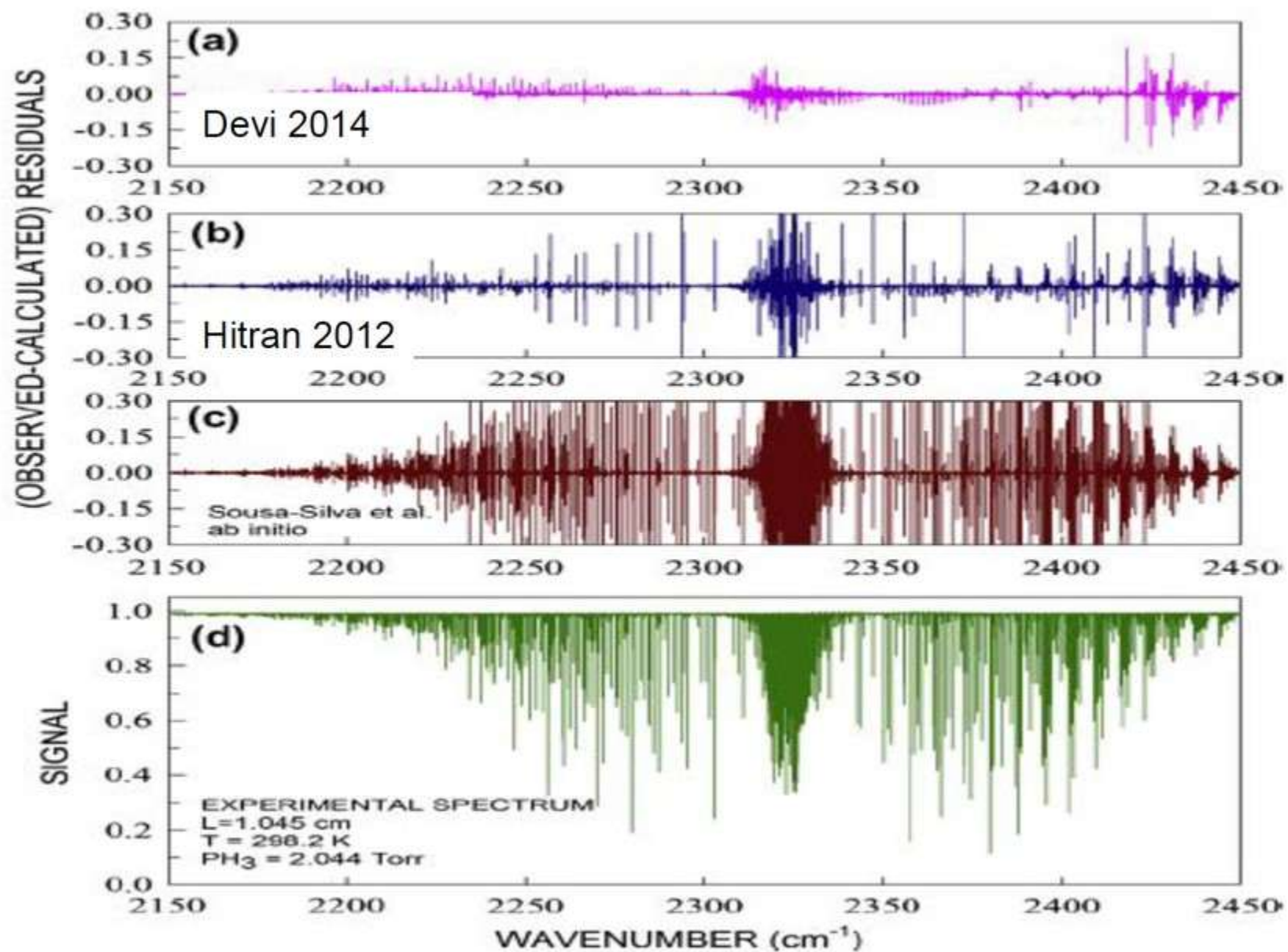
(Fusina and al, J.Mol.Struc.,2000)

-Intensities: rms=**2%** (Brown and al., JMS, 2001)

-Linewidth: self **Broadening** coefficients

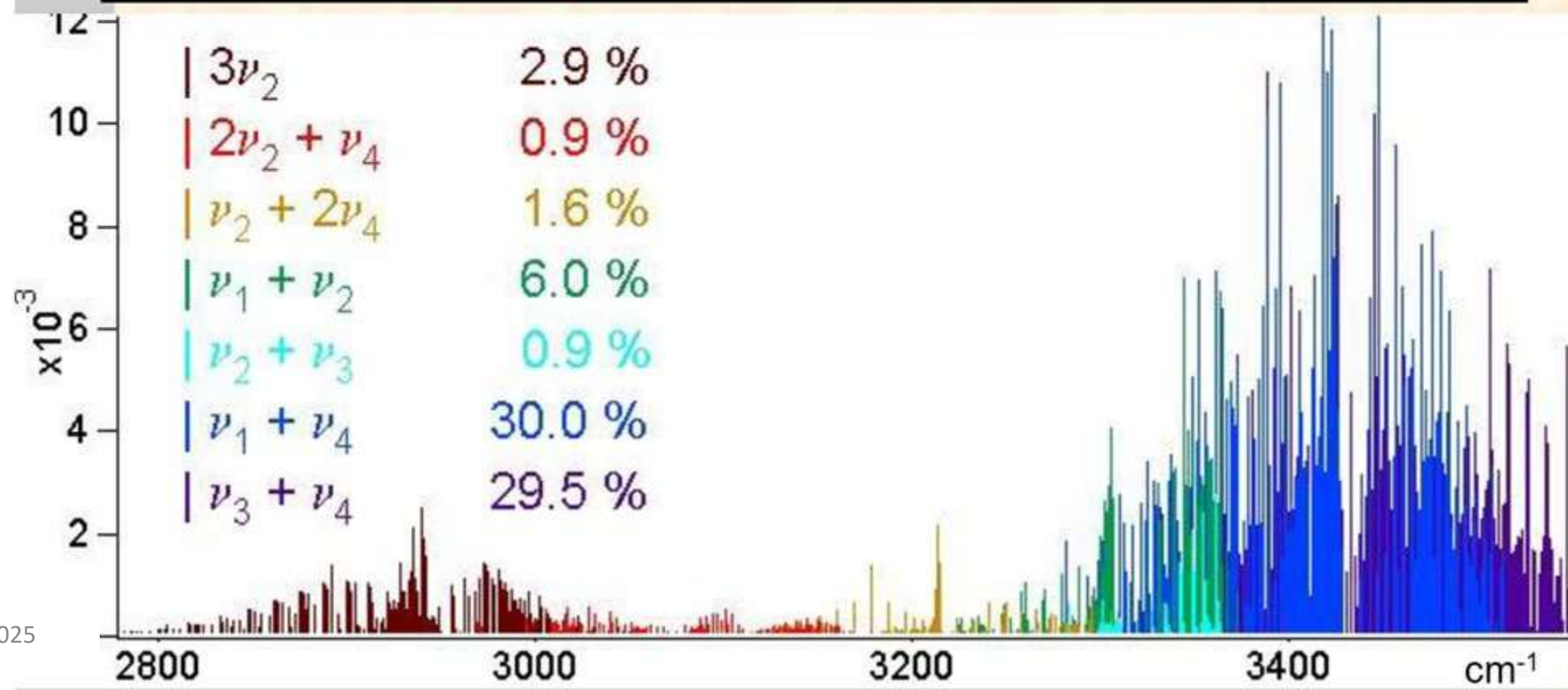
(J. Salem and al, JMS, 2004)

Observed-calculated in the pentad region



ANALYSIS OF THE 3 MICRON REGION :

- Data 2750 - 3600 cm^{-1} : 8075 line positions and intensities obtained by L. R. Brown and R. Butler (Kitt Peak FTS, spectral resolution of 0.0115 cm^{-1}).
- Analysis of spectra: 4095 lines identified.
- 7 vibrational states simultaneously analysed
- **79% of intensity of region accounted for**, - Unassigned state is $3\nu_4$



Isotopes

Isotopes can be treated as regular molecular structures : isotopic ratio have no reason to be similar to Solar System values (Warning : terrestrial databases include the « standard » terrestrial ratio)

Special case : some isotopes change the symmetry of the molecule and can therefore have a very different spectroscopy – ex. CH_3D

Emission and absorption coefficients

Calculation of transition probabilities =>

- * related to the molecular dipole momentum of the molecule
- Depends on the population of the emitting/absorbing level
- Basis : spontaneous, stimulated and absorption coefficients (Einstein coefficients)
- Emission and radiation : Enstein relations A_{ij} , B_{ij} , B_{ji}

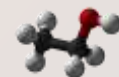
Boltzmann distribution

Spectral absorption coefficient

Types of line broadening

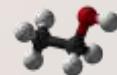
See • Hanson, Summer School on Combustion and the Environment, June 19-24, 2022, Princeton, NJ, USA

<https://cefrc.princeton.edu/sites/g/files/toruqf1071/files/documents/Lecture%20Notes%20-%20Hanson.pdf>



Additional complexities in radiative transfer

1. Far wings effects
2. Collision Induced absorptions



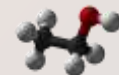
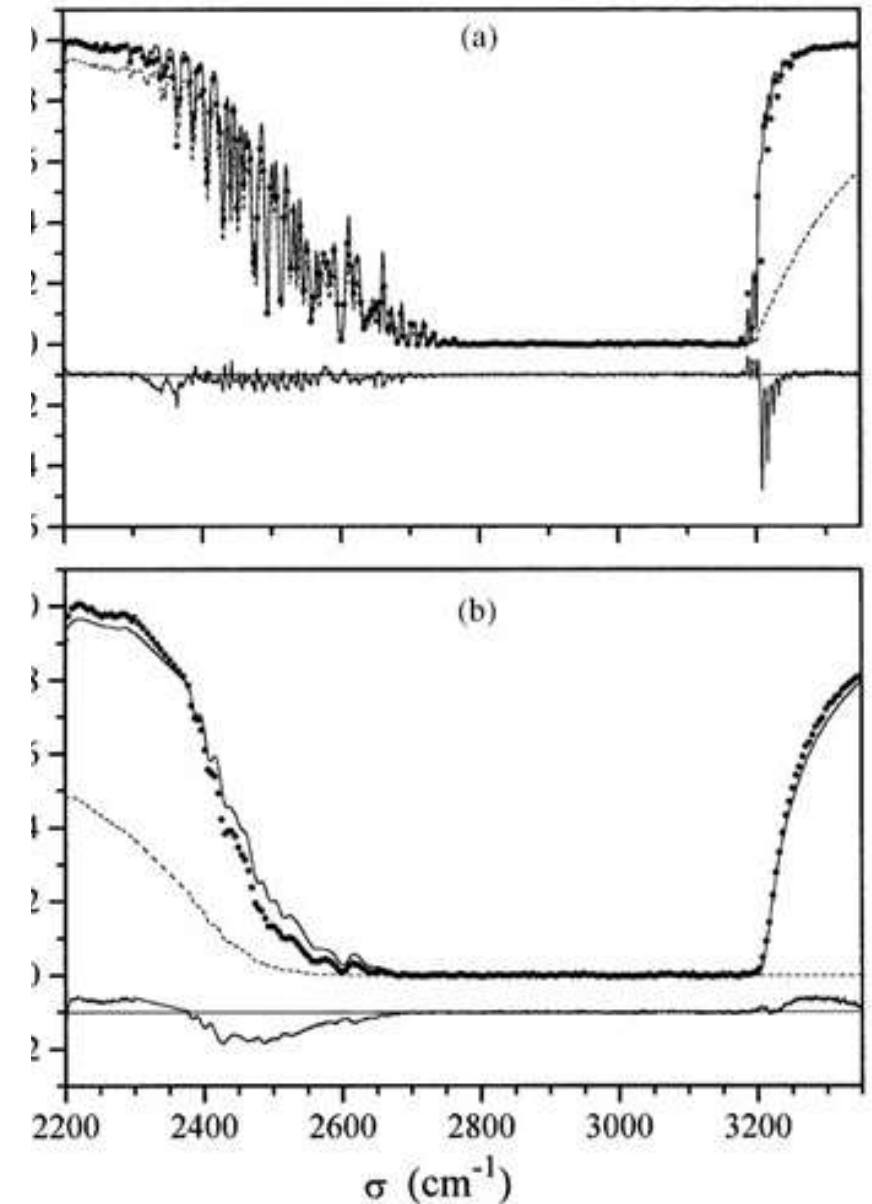
Far wings effects

Very strong lines can dominate the opacity very far from line center: the Lorentz line shape reaches the limit of validity when applied too far from line center

Fig. 1. Transmissions of CH₄-H₂ mixtures at room temperature (297 K) for a 3 m long path. • are measured values. — and - - - have been calculated with the χ factor of Eq. (2) and purely lorentzian line shapes ($\chi=1$), respectively. (a) $d_{\text{CH}_4}=0.919$ Am, $d_{\text{H}_2}=25.0$ Am. (b) $d_{\text{CH}_4}=1.84$ Am, $d_{\text{H}_2}=163$ Am. Measured-calculated residuals are given below the spectra.

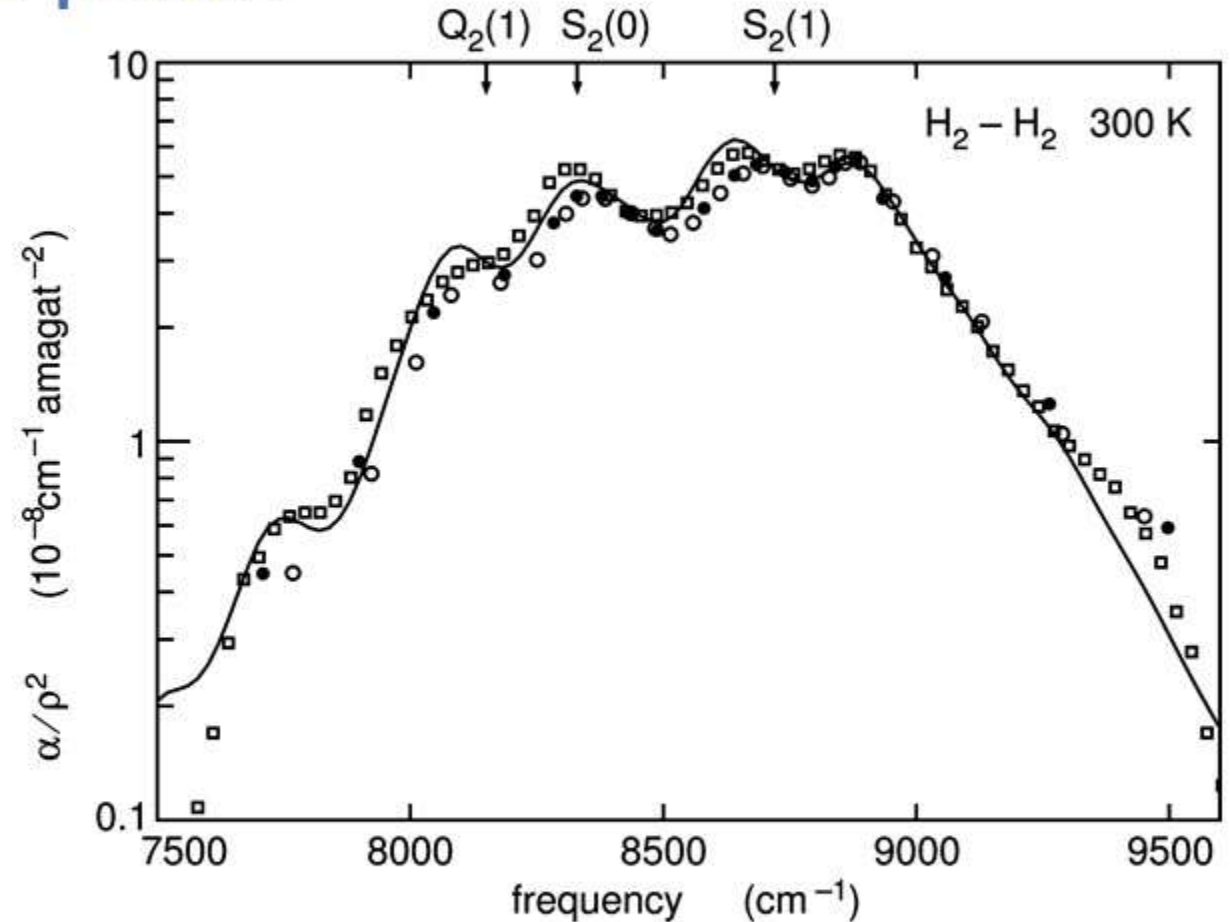
53

Hartmann et al, 2002



Collision induced absorption

The continuum of atmospheric spectra, outside strong molecular bands, is dominated by quadrupolar effects from homopolar diatomic molecules like N_2 or H_2



Abel et al., J. Phys. Chem. A 2011, 115, 25, 6805–6812

Databases: how to compute a spectroscopic database ?

- Example = Hitran extraction
- How to retrieve the database ?
 - Schrödinger equation in Born-Oppenheimer approximation
 - Calculation of a database $\Leftrightarrow$ calculation of the hamiltonian of the molecule
- Practical approaches :
 - combination of prediction/experiments $\Rightarrow$ parametrized hamiltonian
 - Ab initio calculations
 - What is the necessary accuracy for practical use ?